PHYSICAL SCIENCE (EM) EVALIATION KEY INDICATOR

IX – Class

PART -A

1.	Mass percentage = $20/100 \times 100$
	= 20%
2.	Yes.
	O_2 represents Oxygen molecule
	O represents Oxygen atom (1M)
	(OR)
	O ₂ represents a formula
	O represents an atom \int
3.	Carbon
4.	Suspend the object accurately at the centre point.(1M)
	(Any related answer)
5.	u = initial velocity
	v = final velocity
	a = acceleration 2M
	t = time
	(Any related answer 2 M)
6.	m = 300kg
	V = 90km/h = 90 x 5/18 = 25 m/s 1M
	Momentum (P) = mv
	= 300 x 25
	= 7500 kg m/s } ½ M
	(Any related answer)

(OR)

M= 300kg		
V= 90 km/h	½ m	
P= mv		> 2M
=300 x 90	½ m	
= 2700 kg km/h	1M	
i. No changes in mass		
ii. Weight changes	2M	
N-Shell		
2n ²		

9. If symbols were not introduced, it is a problem to write all elements. It is difficult to write chemical equators. Difficult to understand different names at difficult places. (2x1=2)

So Scientists did a good job.

(Any related answer)

10A. i. Consider a circular ring like Bangle. It is a regular shape. It's center of gravity lies at its geometrical center.

Means center of gravity is our side of object.

ii. Consider an arc shaped object like Boomerang. Its center of gravity lies at the centr of its edges along? Straight line. Means Center of gravity is outside of object.

10B.
$$m_1 = 20kg$$

7.

8.

$$M_{1} = 20 kg$$

$$M_{2} = 20 kg$$

$$R = 20 cm = 0.2m$$

$$G = 6.67 x 10^{-11} nm^{2}/kg^{2}$$

$$F = G (m_{1}m_{2}/r^{2})$$

$$I = 6.67 x 10^{-11} x 20 x 20$$

$$\overline{0.2 x 0.2}$$

$$I = 6.67 x 10^{-7} N$$

$$I = 1$$

11.A. Chromatography:

i. Take water in a beaker.

ii. Draw a thick line with marker just above the bottom of the filter paper.

iii. Hang the paper in water by the support of a pencil.

iv. The edge of the paper touches the water. But the marker line.

v. we observe the water gradually moves up and difficult colors up on the paper.

vi. Thus the color components in the into were separated and observed.

11B. i. Prepare a questions solution of lead midrate in a conical task.

ii. Prepare a questions solution of potassium iodide in a test tube.

iii. Hang the test tube in flask without mixing the solutions keep a cork or the flask.

iv. Measure the total mass by using digital balance.

v. Tilt and swirl the flask, so that two solutions mix.

vi. Chemical reaction takes place and new substances formed.

vii. Mea sum the total mass again.

viii. we came to know that the total mass of reactants and products was same.

ix. Means : "The mass neither be created nor destroyed".

"The law of conservation of mass"



12B.

Element	Symbol	Atomic number	Atomic mass	Number of
		(Z)	Number (A)	electrons (N)
Oxygen		8	16	8
Hydrogen		1	1	0
Carbon		6	12	6
Sodium		11	23	12

13A. i. It is a wood machine. – 1M

ii.



iii.







At the point of center of gravity the total man of the body in supposed act at it.4 x 1 = 4M

14 . A	19. A	24. C	29. C
15. B	20. A	25. C	30. A
16. C	21. A	26. A	31. D
17. B	22. A	27. В	32. B
18. C	23. A	28. A	33. D

PART – B

ii.